

# Li-ion battery-charger solutions for JEITA compliance

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## Introduction

Lithium-ion (Li-ion) batteries tend to become dangerous when they are overcharged at high temperatures. Safely charging these batteries has become one of the most important design specifications in battery-powered portable equipment. Progress has been made in establishing industry standards such as the Japan Electronics and Information Technology Industries Association (JEITA) guidelines for improving battery-charging safety. This article addresses safety requirements and battery-charger solutions that meet these requirements in both notebook and single-cell handheld applications.

## Battery-charger safety and the JEITA guidelines

Widely used in consumer electronics from cell phones to laptops, Li-ion batteries have the highest volumetric and gravimetric energy densities among the rechargeable batteries, with no memory effect. They also have a self-discharge rate that is 10 times lower than that of NiMH batteries, and they can provide the instant power required by the system; but are they safe?

Everyone in the industry has seen pictures of exploding laptops and heard about the massive and unprecedented recalls of Li-ion batteries due to cell safety concerns. Such battery explosions or fires originated within the manufacturing process. Batteries contain several metal parts that can sometimes result in undesirable metal impurities within the cell. These impurities are typically sharp metal shards

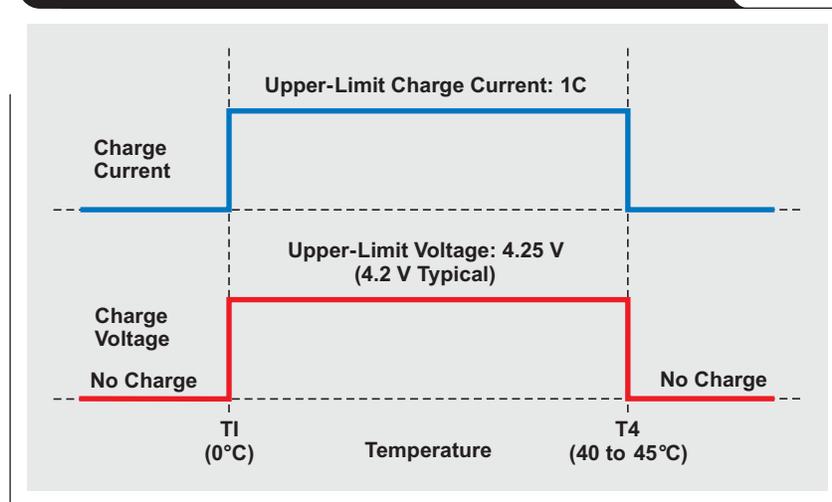
from the battery casing or from electrode materials. If these shards get between the battery's electrode and separator, battery cycling in the negative electrode can eventually cause the shards to puncture the separator. This results in a microshort between the positive and negative electrodes, producing high heat that may ultimately result in fire and/or an explosion.

High temperatures, fire, and explosions are all results of thermal runaway—a condition whereby a battery enters into an uncontrollable reaction. Thermal runaway is a process in which the internal temperature of a battery with  $\text{LiCoO}_2$  as the cathode material and graphite as the anode material reaches approximately  $175^\circ\text{C}$ . This is an irreversible and highly exothermic reaction that can cause a fire, usually when the battery is charging.

Figure 1 shows the charge current and charge voltage over temperature commonly used in the older Li-ion-battery-charging systems that are prone to thermal runaway. Both the battery charge current and charge voltage are constant over the cell temperature from 0 to  $45^\circ\text{C}$ . High cell temperatures not only speed up battery aging but also increase the risk of battery failure.

To improve the safety of charging Li-ion batteries, JEITA and the Battery Association of Japan released new safety guidelines on April 20, 2007. Their guidelines emphasized the importance of avoiding a high charge current and high charge voltage at certain low and high temperature ranges. According to JEITA, problems in the Li-ion batteries occur at high charge voltages and high cell

**Figure 1. Upper-limit charge current and charge voltage in older Li-ion-battery-charging systems**



temperatures. Figure 2 shows the JEITA guidelines for the charge current and charge voltage over cell temperature for batteries used in notebook applications. These batteries have  $\text{LiCoO}_2$  as the cathode active material and graphite as the anode active material.

In the standard charging temperature range from T2 to T3, a Li-ion cell can be charged in the optimal conditions of the upper-limited charge voltage and the upper-limited charge current recommended by the cell's manufacturer for safety.

### Charging at low temperatures

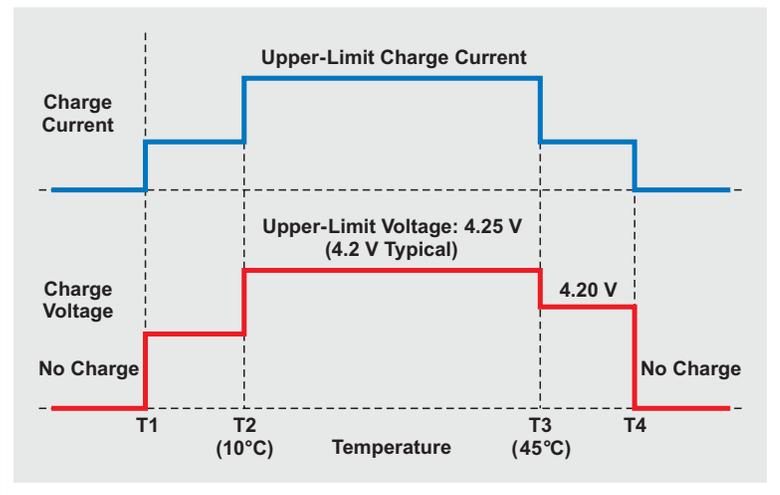
If the cell's surface temperature becomes lower than T2 during charging, the lithium ions could each gain one electron and become metallic lithium. This metallic lithium is likely to deposit on the anode, because at low temperatures the transfer rate decreases and the penetration of lithium ions into the negative electrode carbon slows down. Such metallic lithium could easily react with electrolyte, causing permanent loss of the lithium ions, which degrades the battery faster. In addition, the chemical reaction between metallic lithium and the electrolyte generates a lot of heat, which could lead to thermal runaway. Therefore, the charge current and charge voltage are reduced at low cell temperatures. If the temperature is further reduced to T1 (0°C as an example), the system should not allow charging.

### Charging at high temperatures

If the cell's surface temperature rises above T3 (45°C as an example) during charging, the cathode material,  $\text{LiCoO}_2$ , starts to become more active and can chemically react with the electrolyte when the cell voltage is high. If the cell temperature is further increased to T4, the system should prohibit charging. If the cell temperature reaches 175°C with a cell voltage of 4.3 V, thermal runaway may occur and the battery may explode.

Similarly, Figure 3 shows the JEITA guidelines for charging Li-ion batteries in single-cell handheld applications, where the charge current and charge voltage are also functions of the cell temperature. The maximum charge voltage of 4.25 V includes the battery charger's full tolerance. The battery can be charged at up to 60°C with a reduced charge voltage for safety.

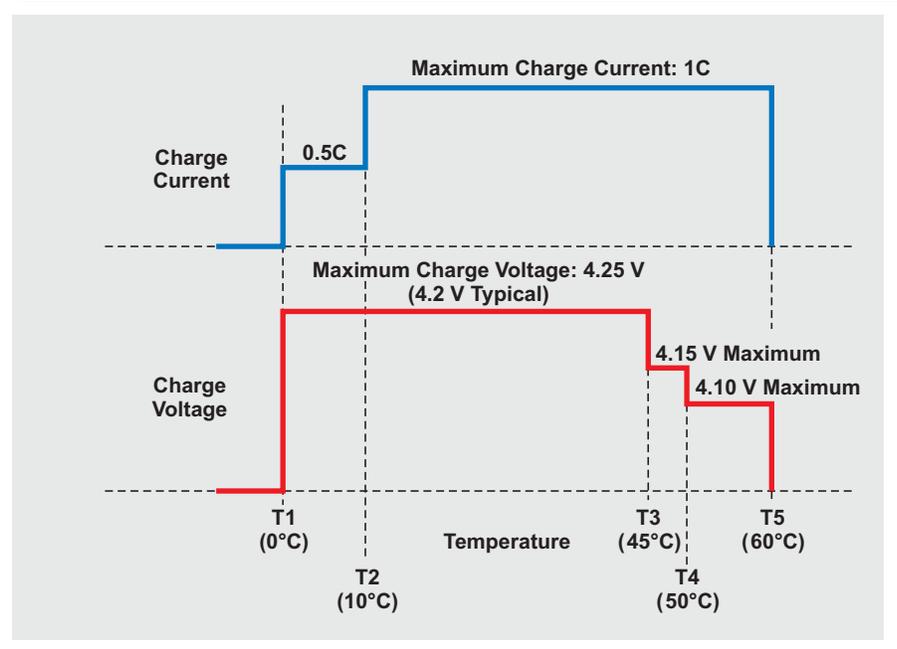
**Figure 2. JEITA guidelines for charging Li-ion batteries in notebook applications**



### Battery-charger solutions for meeting JEITA guidelines

The smart battery pack, which includes a fuel gauge, analog front end, and second-level protector, is commonly used in notebook applications. The fuel gauge provides the battery's cell voltage, charge and discharge current, cell temperature, remaining capacity, and run time to the system through SMBus for optimizing the system performance. The bq20z45 and bq20z40 fuel gauges with Impedance Track™ technology, recently developed by

**Figure 3. JEITA guidelines for charging Li-ion batteries in single-cell handheld applications**





4.06 V when the cell temperature is between 45°C and 60°C. Figure 5 shows a typical application circuit with the bq24050 linear charger. The charger monitors the battery's cell temperature via the thermistor (TS) pin and adjusts the charge current and voltage when the monitored temperature reaches the threshold.

## Conclusion

Charging Li-ion batteries safely is critical and has become one of the key specifications for charger design. Reducing the charge current and voltage at lower and higher

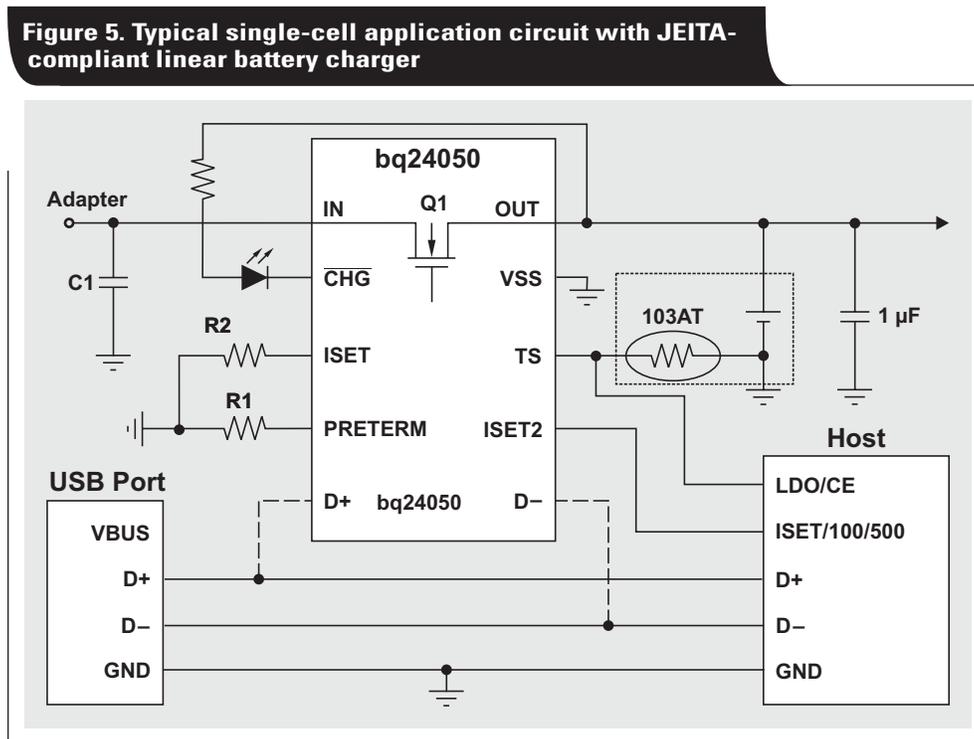
temperature ranges as JEITA recommends can significantly improve the safety of charging these batteries. Both switch-mode and linear battery-charger solutions that comply with JEITA guidelines have been presented.

## Related Web sites

[power.ti.com](http://power.ti.com)

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Replace *partnumber* with bq24050, bq24745, or bq24747



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